

● TOYOBO ENZYMES ●

(Diagnostic Reagent Grade)

# PROTOCATECHUATE 3,4-DIOXYGENASE

*from Pseudomonas sp.*

Protocatechuate:oxygen 3,4-oxidoreductase (deacylizing) (EC 1.13.11.3)



## PREPARATION and SPECIFICATION

Appearance	: Light brown amorphous powder, lyophilized
Activity	: Grade III 3.0U/mg-solid or more (containing approx. 40% of stabilizers)
Contaminant	: NADPH oxidase $\leq 1.0 \times 10^{-1}\%$
Stabilizers	: Sugars

## PROPERTIES

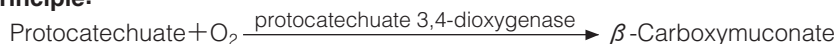
Stability	: Store at $-20^{\circ}\text{C}$ (A decrease in activity of ca. 20% may occur within one year)	(Fig.1)
Molecular weight	: approx. 700,000	
Michaelis constant	: $1.85 \times 10^{-5}\text{M}$ (Protocatechuate)	
Structure	: Protein with nonheme iron	
Inhibitors	: $\text{Ag}^+$ , $\text{Hg}^{++}$ , PCMB	
Optimum pH	: 9.0	(Fig.3)
Optimum temperature	: $60-65^{\circ}\text{C}$	(Fig.4)
pH Stability	: pH 7.0–9.0 (25°C, 72hr)	(Fig.5)
Thermal stability	: below $50^{\circ}\text{C}$ (pH 6.0, 1hr)	(Fig.6)
Effect of various chemicals	: (Table 1)	

## APPLICATIONS

This enzyme is useful for enzymatic determination of choline esterase when coupled with p-hydroxybenzoate hydroxylase.

## ASSAY

### Principle:



The disappearance of protocatechuate is measured at 290nm by spectrophotometry.

### Unit definition:

One unit causes the oxidation of one micromole of protocatechuate per minute under the conditions described below.

### Method:

#### Reagents

- A. Tris-acetate buffer, pH 7.5 : 50mM [Dissolve 6.1g of Tris (MW=121.14) in ca.800ml of H<sub>2</sub>O and, after adjusting pH to 7.5 at 25°C with 0.2M acetic acid, fill up to 1,000ml with H<sub>2</sub>O.]
- B. Protocatechuate acid solution : 0.4mM [Dissolve 6.16mg of protocatechuate in ca.80ml of buffer (A) and, after adjusting pH to 7.5 at 25°C with 1.0N KOH, fill up to 100ml with buffer (A).] (Should be prepared fresh)

#### Procedure

- Pipette 3.0ml of protocatechuate solution (B) into a cuvette (d=1.0cm) and equilibrate at 37°C for about 5 minutes.
- Add 0.05ml of the enzyme solution\* and mix by gentle inversion.
- Record the decrease in optical density at 290nm against water for 3 to 4 minutes in a spectrophotometer thermostated at 37°C, and calculate the  $\Delta\text{OD}$  per minute from the initial linear portion of the curve ( $\Delta\text{OD}$  test).

Concentration in assay mixture	
Tris-acetate buffer	50 mM
Protocatechuate	0.39mM

At the same time, measure the blank rate ( $\Delta\text{OD}$  blank) by using the same method as the test except that the enzyme diluent (A) is added instead of the enzyme solution.

- \* Dissolve the enzyme preparation in ice-cold diluent (A) (1.0mg/ml or more) and dilute to 0.2–0.8U/ml with the same buffer, immediately before assay.

#### Calculation

Activity can be calculated by using the following formula :

$$\text{Volume activity (U/ml)} = \frac{\Delta\text{OD}/\text{min} (\Delta\text{OD test} - \Delta\text{OD blank}) \times V_t \times \text{df}}{3.8 \times 1.0 \times V_s} = \Delta\text{OD}/\text{min} \times 16.1 \times \text{df}$$

$$\text{Weight activity (U/mg)} = (\text{U/ml}) \times 1/C$$

V<sub>t</sub> : Total volume (3.05ml)

V<sub>s</sub> : Sample volume (0.05ml)

3.8 : Millimolar extinction coefficient of protocatechuate (cm<sup>2</sup>/micromole)

1.0 : Light path length (cm)

df : Dilution factor

C : Enzyme concentration in dissolution (c mg/ml)

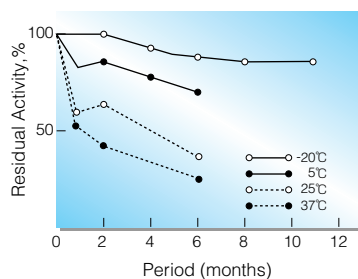
## REFERENCES

- 1) H.Fujisawa and O.Hayashi; *J.Biol.Chem.*, 243, 2673 (1968)

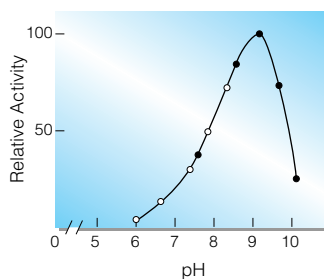
**Table 1. Effect of Various Chemicals on Protocatechuate 3,4-dioxygenase**  
 [The enzyme was incubated with each chemical at 30°C for 1hr. ]

Chemical	Concn.(mM)	Residual activity	Chemical	Concn.(mM)	Residual activity
None	—	100%	PCMB	1.0	0
Metal salt	1.0		NaF	1.0	98
ZnCl <sub>2</sub>		91	NaN <sub>3</sub>	1.0	79
CuSO <sub>4</sub>		90	EDTA	5.0	100
AgNO <sub>3</sub>		0	Borate	50	90
MgSO <sub>4</sub>		95	Tween 20	0.1%	95
BaCl <sub>2</sub>		100	Brij 35	0.1%	112
MnCl <sub>2</sub>		107	SDS	0.05%	85
NiCl <sub>2</sub>		98	Na-cholate	0.1%	93
CaCl <sub>2</sub>		109			
HgCl <sub>2</sub>		3.4			
CrCl <sub>2</sub>		100			
CdCl <sub>2</sub>		100			
CoCl <sub>2</sub>		95			
FeSO <sub>4</sub>		88			

PCMB, p-Chloromercuribenzoate; EDTA, Ethylenediaminetetraacetate; SDS, Sodium dodecyl sulfate

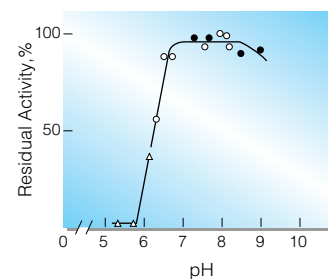


**Fig.1. Stability (Powder form)**  
 [kept under dry conditions ]



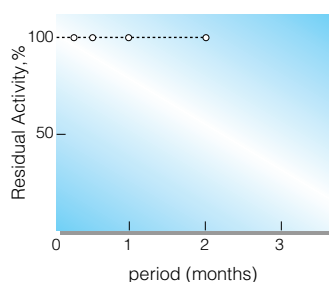
**Fig.3. pH-Activity**

[ 37°C in 50mM buffer solution:  
 pH6.0-8.0, Tris-acetate; pH7.5-10.5,  
 glycine-NaOH ]



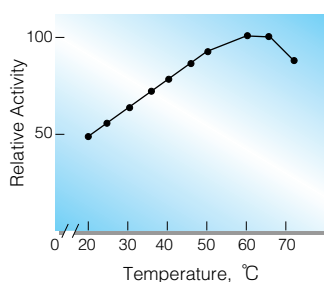
**Fig.5. pH-Stability**

[ 25°C, 72hr-treatment with 50mM  
 buffer solution; pH3.3-5.5,  
 acetate; pH5.5-8.0 K-phosphate;  
 pH6.5-9.0, Tris-acetate ]



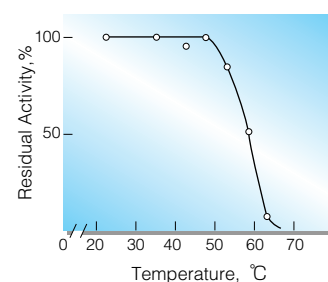
**Fig.2. Stability (Liquid form at 25°C)**

[ enzyme concentraton:35U/ml  
 buffer composition:50mM Tris-HCl buffer, pH7.8 ]  
 contg. 0.1%NaN<sub>3</sub>



**Fig.4. Temperature activity**

[ in 50mM K-phosphate buffer ,pH7.5 ]



**Fig.6. Thermal stability**

[ 1hr-treatment with 50mM K-phosphate ]  
 buffer,pH6.0

## 活性測定法 (Japanese)

### 1. 原理

Protocatechuate + O<sub>2</sub>  $\xrightarrow{\text{protocatechuate 3,4-dioxygenase}}$   
β-Carboxymuconate

プロトカテキュ酸の消失量を290nmの吸光度の変化で測定する。

### 2. 定義

下記条件で1分間に1マイクロモルのプロトカテキュ酸が酸化される酵素量を1単位(U)とする。

### 3. 試薬

- A. 50mM Tris-酢酸緩衝液, pH7.5 [6.1gのトリス (MW=121.14)を約800mlの蒸留水で溶解し, 0.2M酢酸でpH7.5(25°C)に調整後, 蒸留水で1,000mlとする]
- B. 0.4mMプロトカテキュ酸溶液 [6.16mgのプロトカテキュ酸を緩衝液(A)で溶解し, 1N KOHでpH7.5(25°C)に調整後, 緩衝液(A)で100mlとする] (用時調製)

酵素溶液: 酵素標品を予め氷冷した緩衝液Aで溶解 (1.0mg/ml以上)し, 分析直前に同緩衝液で0.2~0.8U/mlに希釈する。

### 4. 手順

- ① 基質溶液(B)3.0mlをキュベット(d=1.0cm)に採り, 37°Cで約5分予備加温する。
- ② 酵素溶液0.05mlを添加し, ゆるやかに混和後, 水を対照に37°Cに制御された分光光度計で290nmの吸光度変化を3~4分間記録し, その初期直線部分から1分間当りの吸光度変化を求める(ΔOD test)。
- ③ 盲検は基質溶液(B)に, 酵素溶液の代わりに酵素希釈液(A)を0.05ml加え, 上記同様に操作を行って, 1分間当りの吸光度変化を求める(ΔOD blank)。

### 5. 計算式

$$U/ml = \frac{\Delta OD/min (\Delta OD_{test} - \Delta OD_{blank}) \times 3.05(ml) \times \text{希釈倍率}}{3.8 \times 1.0 \times 0.05(ml)}$$

$$= \Delta OD/min \times 16.1 \times \text{希釈倍率}$$

$$U/mg = U/ml \times 1 / C$$

3.8 : プロトカテキュ酸のミリモル分子吸光係数 (cm<sup>2</sup>/micromole)

1.0 : 光路長(cm)

C : 溶解時の酵素濃度(c mg/ml)