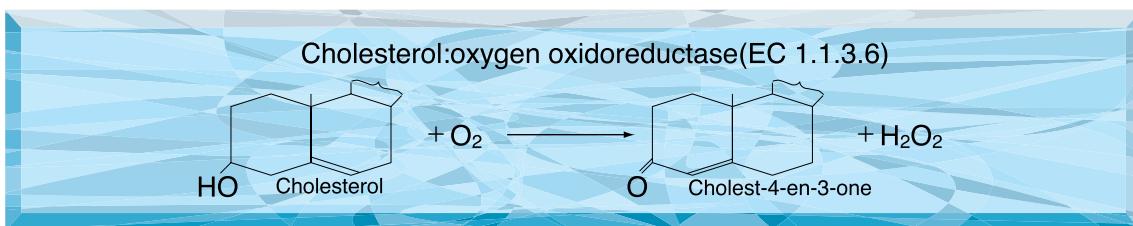


# TOYOBO ENZYMES

## (Diagnostic Reagent Grade)

# **CHOLESTEROL OXIDASE**

*from Microorganism*



## **PREPARATION and SPECIFICATION**

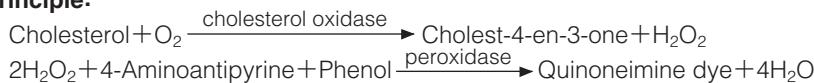
Appearance	: Yellowish amorphous powder, lyophilized
Activity	: Grade III 12U/mg-solid or more
Contaminants	: Catalase $\leq 1.0 \times 10^{-1}\%$ Cholesterol esterase $\leq 1.0 \times 10^{-2}\%$
Stabilizers	: Bovine serum albumin, amino acids

 PROPERTIES

Stability	: Stable at $-20^{\circ}\text{C}$ for at least 9 months	(Fig.1)
Molecular weight	: approx. 55,000 (by gel-filtration)	
Michaelis constants	: $3.0 \times 10^{-5}\text{M}$ (Cholesterol)	
Inhibitors	: Ionic detergents, $\text{Hg}^{++}$	
Optimum pH	: 7.0–8.0	(Fig.3)
Optimum temperature	: $60^{\circ}\text{C}$	(Fig.4)
pH Stability	: pH 5.0–10.0 ( $25^{\circ}\text{C}$ , 20hr)	(Fig.5)
Thermal stability	: below $60^{\circ}\text{C}$ (pH 7.0, 15min)	(Fig.6)
Substrate specificity	: (Table 1)	
Effect of various chemicals	: (Table 2)	

## APPLICATIONS

This enzyme is useful for enzymatic determination of cholesterol in serum when coupled with cholesterol esterase (COE-301, COE-311, COE-313) in clinical analysis.


**ASSAY**
**Principle:**

The appearance of quinoneimine dye formed when coupled with 4-aminoantipyrine and phenol is measured at 500nm by spectrophotometry.

**Unit definition:**

One unit causes the formation of one micromole of hydrogen peroxide (half a micromole of quinoneimine dye) per minute under the conditions described below.

**Method:****Reagents**

A. 0.1M K-Phosphate buffer, pH 7.0

- B. Cholesterol solution : To 5.0ml of Triton X-100 on a hot plate or in a water bath, add 500mg of cholesterol and mix with a stirring bar until cholesterol dissolves. Add 90ml of distilled water to the hot cholesterol-Triton X-100 solution by slowly pouring along a stirring bar. Stir and allow to boil for 30 to 60 seconds. The solution will be cloudy. Cool under running water with gentle agitation, the solution will turn clear. Add 4.0g of sodium cholate and dissolve. Fill up the solution to 100ml with distilled water. This solution is stable for about one week at room temperature. If it becomes cloudy, warm slightly while stirring until it clears.
- C. 4-AA solution : 1.76% (1.76g 4-aminoantipyrine/100ml of H<sub>2</sub>O)
- D. Phenol solution : 6.0% (6.0g phenol/100ml of H<sub>2</sub>O)
- E. POD solution : Horseradish peroxidase 15,000 purpurogallin units/100ml of buffer (A)
- F. Enzyme diluent : 20mM K-Phosphate buffer, pH 7.0 contg.0.2% bovine serum albumin

**Procedure**

1. Prepare the following working solution (20 tests volume), immediately before use and store on ice in a brownish bottle.

51.0ml	Buffer solution	(A)
4.0ml	Substrate solution	(B)
1.0ml	4-AA solution	(C)
2.0ml	POD solution	(E)

2. Pipette 2.9ml of working solution into a cuvette (d=1.0cm) and equilibrate at 37°C for about 3 minutes. Add 0.1ml of Phenol solution (D), mix and keep at 37°C for another 2 minutes.

3. Add 0.1ml of the enzyme solution\* and mix with gentle inversion.

4. Record the increase in optical density at 500nm against water for 3 to 4 minutes in a spectrophotometer thermostated at 37°C, and calculate the Δ OD per minute from the initial portion of the curve (Δ OD test).

At the same time, measure the blank rate (Δ OD blank) by using the same method as the test except that the enzyme diluent is added instead of the enzyme solution.

- \* Dissolve the enzyme preparation in ice-cold enzyme diluent (F), and dilute to 0.1–0.3 U/ml with the same buffer, and store on ice.

Concentration in assay mixture	
K-Phosphate buffer	87 mM
Cholesterol	0.89mM
4-Aminoantipyrine	1.4 mM
Phenol	21 mM
Triton X-100	0.34 %
Sodium cholate	64 mM
BSA	33 μ g/ml
POD	5 U/ml

**Calculation**

Activity can be calculated by using the following formula :

$$\text{Volume activity (U/ml)} = \frac{\Delta \text{OD/min} (\Delta \text{OD test} - \Delta \text{OD blank}) \times Vt \times df}{13.78 \times 1/2 \times 1.0 \times Vs} = \Delta \text{OD/min} \times 4.499 \times df$$

Weight activity (U/mg)=(U/ml) × 1/C

Vt : Total volume (3.1ml)

Vs : Sample volume (0.1ml)

13.78 : Millimolar extinction coefficient of quinoneimine dye under the assay conditions (cm<sup>2</sup>/micromole)

1/2 : Factor based on the fact that one mole of H<sub>2</sub>O<sub>2</sub> produces half a mole of quinoneimine dye.

1.0 : Light path length (cm)

df : Dilution factor

C : Enzyme concentration in dissolution (c mg/ml)


**REFERENCES**

- 1) W.Richmond; *Clin.Chem.*, 19, 1350 (1973).
- 2) H.M.Flegg; *Ann.Clin.Biochem.*, 10, 79 (1973).
- 3) C.C Alain et al; *Clin.Chem.*, 20, 470 (1974).
- 4) P.N.Tarbutton and C.R.Gunter; *Clin.Chem.*, 20, 724 (1974).
- 5) S.Nomoto; *Rinsho Kensa*, 20, 688 (1976).
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- 7) Y.Nishiya et al; *Protein Engng*, 7, 231 (1997)

Table 1. Substrate Specificity of Cholesterol oxidase

Substrate(0.1mM)	Relative activity	Substrate(0.1mM)	Relative activity
Cholesterol	100.0	Ergosterol	44.0
Pregnenolone	60.0	Lanosterol	1.6
$\beta$ -Cholestanol	60.0	Testosterone	1.0
$\beta$ -Sitosterol	120.0	Androsterone	1.5
Stigmasterol	34.0	Dehydroiso-androsterone	15.0

Table 2. Effect of Various Chemicals on Cholesterol oxidase

[The enzyme (1.0U/ml) dissolved in 10mM K-phosphate buffer, pH 7.0 contg. 0.2% BSA was incubated with each chemical at 25°C for 1hr.]

Chemical	Concn.(mM)	Residual activity	Chemical	Concn.(mM)	Residual activity
None	—	100%	NaF	20	100
Metal salt	2.0		NaN <sub>3</sub>	20	100
MgCl <sub>2</sub>		100	EDTA-2Na	5.0	100
CaCl <sub>2</sub>		100	o-Phenanthroline	2.0	100
Ba(OAc) <sub>2</sub>		100	$\alpha,\alpha'$ -Dipyridyl	1.0	100
FeCl <sub>3</sub>		100	Borate	50	100
CoCl <sub>2</sub>		100	IAA	2.0	100
MnCl <sub>2</sub>		100	NEM	2.0	100
Zn(OAc) <sub>2</sub>		100	Hydroxylamine	2.0	100
Cd(OAc) <sub>2</sub>		100	2-Mercaptoethanol	2.0	100
NiCl <sub>2</sub>		100	Triton X-100	0.10%	100
CuSO <sub>4</sub>		90	Tween 20	0.10%	94
Pb(OAc) <sub>2</sub>		100	Span 20	0.10%	90
HgCl <sub>2</sub>	0		Na-cholate	0.10%	100
PCMB	2.0	100	SDS	0.05%	100
MIA	2.0	100	DAC	0.05%	100

Ac, CH<sub>3</sub>CO; PCMB, p-Chloromercuribenzoate; MIA, Monoiodoacetate; NEM, N-Ethylmaleimide; IAA, Iodoacetamide; EDTA, Ethylenediaminetetraacetate; SDS, Sodium dodecyl sulfate; DAC, Dimethyl-benzyl-alkyl-ammonium-chloride.

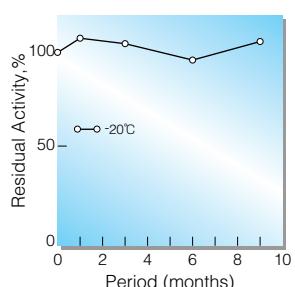


Fig.1. Stability (Powder form)  
[kept under dry conditions]

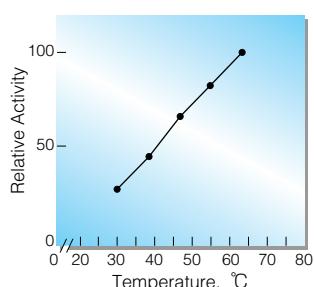


Fig.3. Temperature activity  
[in 0.1M K-phosphate buffer, pH 7.0]

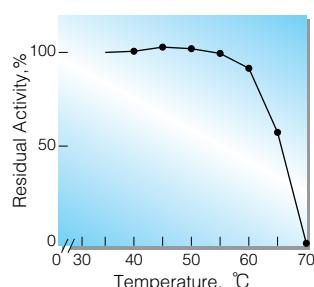


Fig.5. Thermal stability  
[15 min-treatment with 50mM K-phosphate buffer, pH7.0]

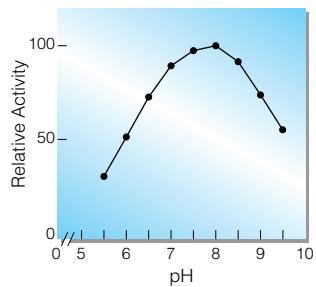


Fig.2. pH-Activity  
[37°C in 0.1M K-phosphate buffer solution]

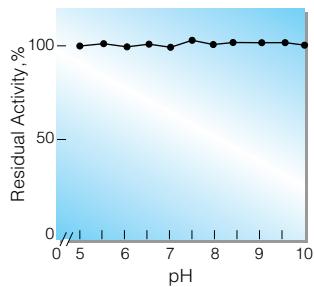


Fig.4. pH-Stability  
[25°C 20 hr-treatment with 50mM buffer solution]  
pH5.0-6.0, acetate buffer  
pH6.5-8.5, K-phosphate buffer  
pH9.0-10.0, Gly-NaOH buffer

## 活性測定法 (Japanese)

### 1. 原理

Cholesterol + O<sub>2</sub>  $\xrightarrow{\text{cholesterol oxidase}}$  Cholest-4-en-3-one + H<sub>2</sub>O<sub>2</sub>  
 2H<sub>2</sub>O<sub>2</sub> + 4-Aminoantipyrine + phenol  $\xrightarrow{\text{peroxidase}}$   
 Quinoneimine dye + 4H<sub>2</sub>O  
 4-Aminoantipyrineとフェノールの酸化縮合生成物であるQuinoneimine色素を500nmで測定し,上記反応で生成したH<sub>2</sub>O<sub>2</sub>量を定量する。

### 2. 定義

下記条件下で1分間に1マイクロモルのH<sub>2</sub>O<sub>2</sub>を生成する酵素量を1単位(U)とする。

### 3. 試薬

- A. 0.1M K-リン酸緩衝液,pH7.0
  - B. コレステロール溶液 [5.0mlのTriton X-100に500mgのコレステロールを添加し,ヒーター上で攪拌溶解する。これに90mlの蒸留水を静かに添加し,攪拌混合後,ヒーター上で30~60秒煮沸する(溶液は濁る)。次いでゆるやかに攪拌しながら流水中で冷却し(溶液は清澄化する),これに4.0gのコール酸ナトリウム塩(ナカライトスク製)を添加して攪拌溶解させた後,蒸留水で最終液量を100mlとする](溶液は室温で少なくとも1週間は保存可能,もし保存中に濁る場合は,攪拌しながら加温清澄化すれば良い)
  - C. 4-AA水溶液:1.76% (4-アミノアンチピリン1.76gを水に溶解して100mlとする)
  - D. フェノール水溶液:6.0% (フェノール6.0gを水に溶解して100mlとする)
  - E. POD溶液:Peroxidase 150mg(100プルプロガリン単位/mg)を100mlの緩衝液(A)に溶解する。
- 酵素溶液:酵素標品を予め氷冷した0.2%のBSAを含む20mM K-リン酸緩衝液, pH7.0で溶解し,同緩衝液で0.1~0.3 U/mlに希釈する。

### 4. 手順

- ①下記反応混液を調製する。
 

51.0ml	K-リン酸緩衝液	(A)
4.0ml	基質溶液	(B)
1.0ml	4-AA水溶液	(C)
2.0ml	POD溶液	(E)

(褐色瓶にて氷冷保存)
- ②反応混液2.9mlをキュベット(d=1.0cm)にとり,37°Cで約3分間予備加温し,0.1mlのフェノール水溶液を加えて更に2分間加温する。
- ③酵素溶液0.1mlを加え,ゆるやかに混和し,水を対照に37°Cに制御された分光光度計で500nmの吸光度の増加を3~4分間記録し,その直線部分から1分間あたりの吸光度変化を求める( $\Delta OD_{\text{test}}$ )。
- ④盲検は反応混液に,酵素溶液の代りに酵素希釈液(0.2% BSAを含む20mM K-リン酸緩衝液,pH7.0)を0.1ml加え,上記同様に操作を行って1分間当たりの吸光度変化を求める( $\Delta OD_{\text{blank}}$ )。

### 5. 計算式

$$\begin{aligned}
 \text{U/ml} &= \frac{\Delta OD/\text{min} (\Delta OD_{\text{test}} - \Delta OD_{\text{blank}}) \times 3.1(\text{ml}) \times \text{希釈倍率}}{13.78 \times 1/2 \times 1.0 \times 0.1(\text{ml})} \\
 &= \Delta OD/\text{min} \times 4.499 \times \text{希釈倍率} \\
 \text{U/mg} &= \text{U/ml} \times 1 / C \\
 13.78 &: \text{Quinoneimine色素の上記測定条件下でのミリモル分子吸光係数(cm}^{\text{2}}/\text{micromole}) \\
 1/2 &: \text{酸素反応で生成したH}_2\text{O}_2\text{の2分子のから形成するQuinoneimine色素は1分子である事による係数。} \\
 1.0 &: \text{光路長(cm)} \\
 C &: \text{溶解時の酵素濃度(c mg/ml)}
 \end{aligned}$$